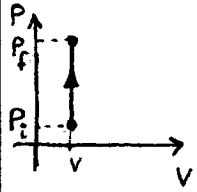
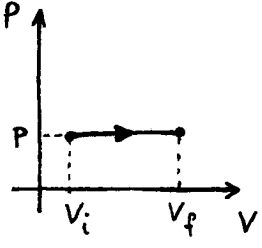
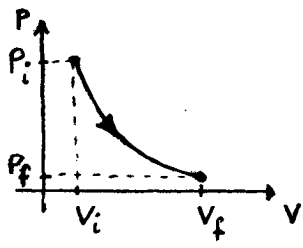
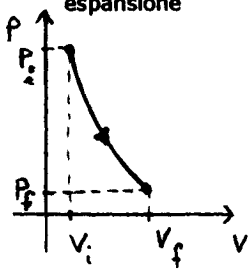


Calore, Lavoro e variazione di Energia Interna nelle principali trasformazioni

	Isocora	Isobara	Isoterma	Adiabatica
Legge	$\frac{P_A}{T_A} = \frac{P_B}{T_B}$ Gay - Lussac	$\frac{V_A}{T_A} = \frac{V_B}{T_B}$ Gay - Lussac	$P_A \cdot V_A = P_B \cdot V_B$ Boyle	$P_A \cdot V_A^\gamma = P_B \cdot V_B^\gamma$ $T_A \cdot V_A^{\gamma-1} = T_B \cdot V_B^{\gamma-1}$
	$\Delta V=0 \rightarrow L=0$		$\Delta T=0 \rightarrow \Delta U=0$	$Q=0$
ΔU	$\Delta U=Q$	$\Delta U=Q-L$	$\Delta U=0$	$\Delta U=-L$
Q	$Q=nC_V \Delta T$	$Q=nC_p \Delta T$	$Q=L$	$Q=0$
L	$L=0$	$L=P \cdot \Delta V$	$L = P_A V_A \ln \frac{V_B}{V_A}$ $L = P_A V_A \ln \frac{P_A}{P_B}$	$L = nC_V (T_1 - T_2)$
	$P_f > P_i \rightarrow Q +$ Volume bloccato	$V_f > V_i \rightarrow Q +, L +$ espansione	$V_f > V_i \rightarrow Q +, L +$ espansione	$V_f > V_i \rightarrow L +$ espansione
				
	$P_f < P_i \rightarrow Q -$ Volume bloccato	$V_f < V_i \rightarrow Q -, L -$ compressione	$V_f < V_i \rightarrow Q -, L -$ compressione	$V_f < V_i \rightarrow L -$ compressione
	